文章编号: 1001-6880(2014) 2-0155-04

# 瑞香狼毒根中的一个新化合物

杨彩霞123\* 雷 蕾1 王小亮1 张正凯1

1西北师范大学化学化工学院:

2甘肃省高分子材料重点实验室;3生态环境相关高分子材料教育部重点实验室 兰州 730070

摘 要: 从瑞香狼毒根中分离得到了 7 个化合物 根据理化性质和波谱数据鉴定为: 3-丁酰基 4-氨基肉桂酸乙酯 (1)、阿魏酸 (2)、香草酸 (3)、姜黄素 (4)、5 (4) 、5 (4) 、5 (4) 、5 (4) 、5 (4) 、5 (4) 、5 (4) 、6 (4) 、7 (4) 、6 (4) 、7 (4) 、6 (4) 、7 (4) 、6 (4) 、7 (4) 、6 (4) 、8 (4) 、8 (4) 、9 (4)

关键词: 瑞香狼毒; 瑞香科; 酚性化合物; 苯丙烯化合物

中图分类号: R284. 2

文献标识码: A

## A New Compound Isolated from Roots of Stellera chamaejasme L.

YANG Cai-xia<sup>1 2 3\*</sup> LEI Lei<sup>1</sup> ,WANG Xiao-liang<sup>1</sup> ZHANG Zheng-kai <sup>1</sup>

<sup>1</sup>College of Chemistry and Chemical Engineering Northwest Normal University, Key Laboratory polymer Materials of Gansu Province, <sup>3</sup>Key Laboratory of Eco-environment Related Polymer Materials Ministry of Education Lanzhou 730070 China

Abstract: Seven compounds were isolated from the roots of *Stellera chamaejasme* L. Their structures were identified as 3-butyryl-4-amino ethylcinnamate(1) ferulic acid(2) vanillic acid(3) daphneticin(4) 5´-demethoxy daphneticin(5), 3´-hydroxy-4´-O-β-D-glucopyranoside flavone(6) and 3-methoxy-4-O-β-D-gluco-benzoic acid(7) by comparing physicochemical properties and NMR data with published literatures. Among them compounds 1-3 6-7 were isolated from this species for the first time compound 1 is a new compound.

Key words: Stellera chamaejasme L.; Thymelaeaceae; phenolic compounds; phenylpropanoid

# Introduction

Stellera chamaejasme L. is one species of the genus Stellera belonging to family Thymelaeaceae. In China , this plant is mainly distributed in northwest and northeast areas ,including Inner Mongolia ,Qinghai ,Gansu , Tibet provinces [1]. The dried roots of S. chamaejasme L. is a well-known traditional Chinese medicine. It is often used as expectorant anti-inflammation detoxification apocatastasis agent and also used for the treatment of furuncle carbuncle and ulcers [2] 3]. Zhou L et al [4]. and Gong XX et al [5]. reported the antibacterial ,anti-virus anticancer activity and bactericidal effect of compounds from the roots of S. chamaejasme L. Previous phytochemical investigation on S. chamaejasme L. had resulted in the isolation and identification of a variety

Received: November 18 2013 Accepted: January 5 2014 Foundation item: Foundation for young teachers of Northwest normal university( NWNU-QN-07-44); Startup project of Doctor scientific research of Northwest normal university( NWNU-179)

\* Corresponding author Tel: 86-931-7971533; E-mail: jltycx@ 126. com

of secondary metabolites , which mainly included coumarins ,flavonoids ,lignans and phenylpropanoid glycosides [3-6]. Our laboratory had recently reported a number of coumarins and flavonoids from the roots of S. chamaejasme L. [7]. As a continuation of our phytochemical investigation ,the present study reported another 7 compounds of this plant. The structures of these compounds were identified as 3-butyryl-4-amino ethylcinnamate (1), ferulic acid (2), vanillic acid (3), daphneticin (4) ,5'- demethoxy daphneticin (5) ,3'hydroxy-4'-0-\beta-D-glucopyranoside flavone (6) and 3methoxy-4-O-\beta-D-gluco-benzoic acid(7) by comparing physicochemical properties and NMR data with published literatures. Among them, compounds 1-3,6-7 were reported from this species for the first time compound 1 is a new compound.

### **Materials and Reagents**

The roots of *S. chamaejasme* L. were collected in August 2008 from Huining county of Gansu province China. It

was identified by Professor CHEN Xue-lin (College of Life Science Northwest Normal University). A voucher specimen (No. 200807) was deposited in the College of Chemistry and Chemical Engineering, Northwest Normal University.

 $^1H$  NMR and  $^{13}C$  NMR spectral data were recorded on a Bruker-DRX-400 FT NMR spectrometer ( 400 MHz for  $^1H$  NMR and 100 MHz for  $^{13}C$  NMR) with tetramethylsilane (TMS) as internal standard; Electron ionization mass spectrometry (EI-MS) and HREI-MS spectral data were acquired on a Bruker APEX II; Silica gel ( 200-300 300-400 mesh) and silica gel GF $_{254}$  ( 10-40  $\mu$ m) were purchased from Qingdao Hai Yang Chemical Group Company ,Shandong; Sephadex LH-20 gel were used for column chromatography; Spots were detected on TLC under UV lamp or by heating after spraying with 5%  $H_2SO_4$  in  $C_2H_5OH(\,v/v)$  .

#### **Extraction and Isolation**

The powder of dried roots (5 kg) of S. chamaejasme L. was exhaustively extracted with 90% ethanol 3 times under reflux. The combined extract was evaporated under reduced pressure to yield a syrupy residue (670 g) . The residue was suspended in water and extracted with petroleum ether (PE ,60-90 °C) ,CHCl<sub>2</sub> ,EtOAc and n-BuOH successively. The PE fraction (71 g) CHCl<sub>3</sub> fraction (26 g) ,EtOAc fraction (125 g) and n-BuOH fraction (54 g) were yielded respectively. The CHCl<sub>3</sub> fraction was chromatographed on silica gel column using gradient elution with PE/CHCl<sub>3</sub> (50:1 to 1 :1) to yield 6 fractions (Fr. 1-Fr. 6). Fr. 5 was sequentially separated on silica gel column eluted with CHCl<sub>3</sub>/CH<sub>3</sub>COCH<sub>3</sub>(30:1 to 1:1) to yield 2(20.9) mg) and 3(9.0 mg). The EtOAc fraction was subjected to silica gel column gradiently eluted with CHCl<sub>3</sub>/ EtOAc(50:1 to 5:1) ,18 fractions were obtained. Fr. 1 was retreated on silica gel column and eluted with CHCl<sub>3</sub>/EtOAc(20:1 to 10:1) to yield 4(23.3 mg) and 5(12 mg). The n-BuOH fraction was subjected to silica gel column using a gradient of EtOAc/MeOH(30 :1 to 1:1) A fractions were obtained. Fr. 1 was isolated and purified in combination of silica gel and Sephadex LH-20 column to yeild 1 (15 mg). By the same

method compound **6**(14 mg) and **7**(9 mg) were obtained from Fr. 2 eluted with CHCl<sub>2</sub>/MeOH(30:1 to 1:1).

#### Structural elucidation

Compound 1 was obtained as white crystal with a molecular formula of C<sub>15</sub>H<sub>19</sub>O<sub>3</sub>N deduced from its positive HREI-MS data( [M]  $^{+}$  at m/z: 261. 1357 ,calc. 261. 1360). The IR(KBr) spectrum of 1 revealed absorption bands of amino (3480 3325 cm<sup>-1</sup>) carbonyl (1688 cm<sup>-1</sup>), aromatic ring (1603,1510,1444 cm<sup>-1</sup>) and double bond (1661 cm<sup>-1</sup>). The <sup>1</sup>H NMR spectrum of 1 (Table 1) indicated the signals of a 1 3 4-trisubstituted benzene ring at  $\delta$  6.88 7.21 (each 1H  $_{\star}$ d  $_{\star}J$  = 8.0 Hz) and 7.61(1H s), a trans-double bond at  $\delta$  7.58 and 6.27 (each 1H ,d J = 16.0 Hz) ,which suggested the presence of phenylpropanoid moiety [8]. Besides, conspicuous signals of an ethyoxyl group at  $\delta$  4. 19 (2H J = 7.6 Hz) and 1.35 (3H J = 7.6 Hz) ,anactive hydrogen at 3.89 (which can be exchanged by D<sub>2</sub>O) were also observed. The <sup>13</sup>C NMR and DEPT data of 1 (Table 1) contained 15 signals including 10 sp<sup>2</sup> carbons and 5 sp<sup>3</sup> carbons which revealed two carbonyl groups at  $\delta_c$  167.9(s) and 196.9(s), a double bond carbon at 144.8(d) ,114.4(d) ,an aromatic carbons at 124.6(s) ,127.8(d) ,126.2(s) ,145.7(s) ,115.5 (d) ,132.4(d) and an ethyoxyl carbons at 61.4(t) 14.5(q). The presence of a butyryl group was deduced from the signals of  $\delta_{\rm H}$  0.87(t,3H),1.53(m,2H), 2. 88( t ,2H) and  $\delta_{\rm C}$  14. 1 ,18. 7 ,40. 3 and 196. 9. Considering the molecular formula of 1 an ethyoxyl and an amino were presented in 1 in addition to the phenylpropanoid moiety and an butyryl group. The configuration between C-7 and C-8 was trans inferred from the coupling constant (16.0 Hz). The location of butyryl group was assigned by HMBC correlations of  $\delta_{\rm H}$  7. 61 (H-2) with  $\delta_{\rm C}$  196.9(C-1') and  $\delta_{\rm H}$  2.88(H-2') with  $\delta_{\rm C}$  196. 9( C-1') ( Fig. 1) . Besides ,the observed correlations of  $\delta_{\rm H}$  7.58( H-7) with  $\delta_{\rm C}$  124.6( C-1)  $\delta_{\rm H}$  6.27

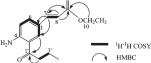


Fig. 1 Key <sup>1</sup>H-<sup>1</sup>H COSY and HMBC correlations of compound 1

(H-8) with  $\delta_{\rm C}$  167.9(C-9) and  $\delta_{\rm H}$  4.19(H-10) with  $\delta_{\rm C}$  167.9(C-9) indicated that **1** had an acryloyl group located at C-1. Analysis and comparison of the  $^{1}$ H- $^{1}$ H COSY and HMBC spectra supported the  $^{1}$ H NMR and  $^{13}$ C NMR assignments (Table 1) and further suggested

that compound 1 was a derivative of phenylpropanoid containing a nitrogen atom. All spectra data of 1 were in agreement with the structure as shown in Fig. 1. Therefore the structure of 1 was established as 3-butyryl-4-amino ethylcinnamate.

Table 1 <sup>1</sup>H NMR(400 MHz) and <sup>13</sup>C NMR(100 MHz) data of 1(in CD<sub>3</sub>OD-d<sub>4</sub>  $\beta$  ppm J in Hz)

Position	$^{1}\mathrm{H}$	<sup>13</sup> C	Position	<sup>1</sup> H	<sup>13</sup> C
1		124.6( s)	9		167.9( s)
2	7.61(s ,1H)	127.8(d)	10	4.19( q 2H $J = 7.6$ )	61.4( t)
3		126.2(s)	11	1.35( t 3H $J = 7.6$ )	14.5( q)
4		145.7(s)	1′		196.9(s)
5	145.7( s)	115.5( d)	2	2.88( t 2H)	40.3(t)
6	7.21(d,1H, $J=8.0$ )	132.4( d)	3	1.53( m 2H)	18.7( t)
7	7.58( d ,1 H $J = 16.0$ )	144.8( d)	4′	0.87( t 3H)	14.1( q)
8	6.27( d ,1 H $J = 16.0$ )	114.4( d)	$\mathrm{NH}_2$	3.89(s)	

**Compound 2** white crystal ,C<sub>10</sub> H<sub>10</sub> O<sub>4</sub> ,EI-MS m/z: 194 [M]<sup>+</sup>; <sup>1</sup>H NMR ( 400 MHz ,CD<sub>3</sub>OD) δ: 7. 19 ( 1H ,d , J = 2. 0 Hz ,H-2) ,7. 08 ( 1H ,dd , J = 2. 0 ,8. 0 Hz ,H-5) ,6. 83 ( 1H ,d ,J = 8. 0 Hz ,H-6) ,7. 63 ( 1H ,d ,J = 16. 0 Hz ,H-7) 6. 34 ( 1H ,d ,J = 16. 0 Hz ,H-8) ,3. 91 ( 3H ,s ,3-OCH<sub>3</sub>); <sup>13</sup> C NMR ( 100 MHz , CD<sub>3</sub>OD) δ: 171. 2 ( COOH) ,127. 8 ( C-1) ,111. 7 ( C-2) ,150. 6 ( C-3) ,149. 3 ( C-4) ,116. 7 ( C-5) ,124. 2 ( C-6) ,146. 8 ( C-7) ,116. 2 ( C-8) ,55. 9 ( 3-OCH<sub>3</sub>) . These spectral data were identical with those of ferulic acid<sup>[9]</sup>.

**Compound 3** white powder ,C<sub>8</sub>H<sub>8</sub>O<sub>4</sub> ,EI-MS m/z: 168 [M]<sup>+</sup>; <sup>1</sup>H NMR ( 400 MHz ,DMSO- $d_6$  ) δ: 7. 43 ( 1H ,d , J = 1. 6 Hz ,H-2 ) β. 84 ( 1H ,d , J = 8. 0 Hz , H-5 ) 7. 44 ( 1H ,dd ,J = 1. 6 ,8. 0 Hz ,H-6 ) ,12. 47 ( 1H ,brs ,-COOH ) 9. 80 ( 1H ,brs ,4-OH ) 3. 81 ( 3H , s 3-OCH<sub>3</sub> ); <sup>13</sup>C NMR ( 100 MHz ,DMSO- $d_6$  ) δ: 167. 2 ( -COOH ) ,121. 7 ( C-1 ) ,115. 1 ( C-2 ) ,147. 4 ( C-3 ) , 149. 8 ( C-4 ) ,112. 7 ( C-5 ) ,123. 6 ( C-6 ) ,56. 3 ( 3-OCH<sub>3</sub> ) . The above spectral data were identical with those of vanillic acid<sup>[10]</sup>.

**Compound 4** colorless needles  $\mathcal{C}_{19}H_{12}O_7$  ,EI-ME m/z: 352 [M]<sup>+</sup>; <sup>1</sup> H NMR ( 400 MHz ,DMSO- $d_6$ ) : 6. 34 ( 1H ,d , J = 9. 6 Hz ,H-3) 8. 00 ( 1H ,d , J = 9. 6 Hz ,H-4) ,7. 20 ( 1H ,d , J = 8. 6 Hz ,H-5) 6. 96 ( 1H ,d , J = 8. 6 Hz ,H-6) 6. 76 ( 2H ,s ,H-2′ ,6′) 4. 43 ( 1H ,d ,

 $J=7.8~{\rm Hz}$  ,H-7′) ,4. 34 ( 1H ,m ,H-8′) ,3. 40 ,3. 68 ( each 1H ,m ,H-9′) ,3. 81 ( 6H ,s ,2 × OCH<sub>3</sub>) ,8. 54 ( 1H ,s ,4′-OH) ,4. 36 ( 1H ,s ,9′-OH) ;  $^{13}$  C NMR ( 100 MHz ,DMSO- $d_6$ )  $\delta$ : 160. 3 ( C-2) ,113. 5 ( C-3) ,143. 5 ( C-4) ,112. 0 ( C-5) ,113. 2 ( C-6) ,147. 7 ( C-7) ,138. 4 ( C-8) ,149. 2 ( C-9) ,113. 6 ( C-10) ,126. 5 ( C-1′) ,106. 1 ( C-2′) ,149. 3 ( C-3′) ,132. 2 ( C-4′) ,149. 3 ( C-5′) ,106. 1 ( C-6′) ,77. 2 ( C-7′) ,78. 5 ( C-8′) ,60. 5 ( C-9′) ,56. 7 ( OCH<sub>3</sub>) . The above spectral data were identical with those of daphneticin [11].

**Compound 5** yellow powder  $C_{19}H_{16}O_7$ , EI-MS m/z: 356 [M]<sup>+</sup>; <sup>1</sup> H NMR (400 MHz ,DMSO- $d_6$ )  $\delta$ : 6. 32 (1 H ,d, J = 10.0 Hz, H--3), 7.98(1 H ,d, J = 10.0)Hz ,H-4) 7. 19(1H ,d , J = 8.0 Hz ,H-5) 6. 95(1H , d, J = 8.0 Hz, H-6) 7.03(1 H, J = 2.0 Hz, H-2), 6. 81 (1H, d, J = 8.0 Hz, H-5'), 6. 88 (1H, dd, J =8. 0 2. 0 Hz ,H-6') 3. 38 3. 70 (1H ,m ,H-9') 3. 80 (3H s -OCH<sub>3</sub>) 4.31(1H ,m ,H-8') 9.19(1H s 4'-OH) 4. 45(1H ,s 9'-OH); 13 C NMR(100 MHz ,DM- $SO-d_6$ )  $\delta$ : 160. 2 ( C-2) ,112. 5 ( C-3) ,144. 6 ( C-4) , 119. 8 ( C-5 ) ,113. 2 ( C-6 ) ,146. 8 ( C-7 ) ,131. 0 ( C-8) ,143. 1 ( C-9 ) ,112. 9 ( C-10 ) ,126. 5 ( C-1' ) ,112. 0 (C-2'), 146. 9(C-3'), 147. 5(C-4'), 115. 3(C-5'), 120. 5 ( C-6') ,76. 5 ( C-7') ,78. 1 ( C-8') ,59. 9 ( C-9') 55.9(-OCH<sub>3</sub>). The above spectral data were identical with those of 5'-demethoxy daphneticin<sup>[11]</sup>.

**Compound 6** colorless needles  $\mathcal{L}_{21}H_{20}O_9$  EI-MS m/z( neg. ): 415 [M-H]<sup>-</sup>; <sup>1</sup>H NMR ( 400 MHz ,DMSO $d_6$ )  $\delta$ : 6. 89(1H s ,H-3) 8. 02(1H ,d ,J = 8. 0 Hz ,H-5) 7.48(1 H, t, J = 7.0 Hz, H-6) 7.80(1 H, m, H-7) 7.79(1H ,m ,H-8) 7.55(1H ,d , J = 2.0 Hz ,H-2') 7. 24(1H ,t ,J = 8.5 Hz ,H-5') 7. 55(1H ,t ,J =8. 5 Hz ,H-6') 9. 10(1H ,brs 3'-OH) 4. 85(1H ,d ,J = 7. 2 Hz ,H-1'') ,3. 15-3. 73 (6H ,protons of glycoside);  $^{13}$  C NMR (100 MHz ,DMSO- $d_6$ )  $\delta$ : 162. 5 (C-2) ,105. 7 ( C-3) ,177. 2 ( C-4) ,125. 6 ( C-5) ,134. 4 (C-6), 118. 7 (C-7), 124. 6 (C-8), 156. 9 (C-9), 124. 9( C-l0) ,123. 5( C-l') ,113. 5( C-l') ,146. 9( C-3') ,148.5 ( C-4') ,115.8 ( C-5') ,118.7 ( C-6') , 101. 3( C-1 ") ,73. 3( C-2 ") ,77. 4( C-3 ") ,69. 7( C-4") 75.8(C-5") 60.8(C-6"). These spectral data were identical with those of 3'-hydroxy -4'-O-β-D-glucopyranoside flavone<sup>[12]</sup>.

Compound 7 white powder ,C<sub>14</sub> H<sub>18</sub> O<sub>9</sub> ,EI-MS m/z: 330 [M]<sup>+</sup>; <sup>1</sup>H NMR ( 400 MHz ,DMSO- $d_6$ ) δ: 7. 74 ( 1H ,brs ,H-2) ,7. 48 ( 1H ,d , J = 8. 4 Hz ,H-6) ,7. 14 ( 1H ,d J = 8. 4 Hz ,H-5) 5. 01 ( 1H ,d J = 6. 0 Hz ,H-1′) 3. 66 ( 1H ,d J = 12. 0 Hz H-6′a) 3. 45 ( 1H ,dd ,J = 5. 6 ,12. 0 Hz ,H-6′b) 3. 35 ( 1H ,m ,H-2′) 3. 18 ( 1H ,m ,H-4′) ,3. 35 ( 1H ,m ,H-5′) ,3. 80 ( 3H ,s ,3-0CH<sub>3</sub>) . These spectral data were identical with those of 3-methoxy-4-O- $\beta$ -D-gluco-benzoic acid [13] .

#### References

- Editorial Board of Chinese Florae of the Chinese Academy of Science. Flora of China, Beijing: Science Press, 1999. 372– 374.
- 2 Jiangsu New Medical College. Dictionary of Chinese Material

- Medic. Shanghai: Shanghai People's Publishing House ,1986. 1977
- 3 Pei YH ,Fen BM ,Hua HM *et al.* Advances in chemical and pharmacological studies on *Stellera chamaejasme*. *Chin Tradit Herb Drugs* 2001 32:764-766.
- 4 Zhou L ,Yuan C ,Qin BF et al. Study on the antimicrobial activity contituents from the roots of Stellera chamaejasme L. .
  ( II ) . Acta Bot Borea-occident Sin 2004 24: 2346-2349.
- 5 Gong XX ,Li WJ ,Ouyang Q et al. Study on the antibacterial substances from the root of Stellera chamaejasme L. . J Sichuan Uni 2006 43:697-701.
- 6 Yu BQ ,Xu WC. Advances chemical constituents and activity studies on *Stellera chamaejasmme* L. . *Agrochemicals* ,2008 , 47: 863-866.
- 7 Yang CX, Wei B, Wang FP, et al. Flavonoids from the roots of Stellera chamaejasme L. Nat Prod Res Dev(天然产物研究与开发) 2012 24:1374-1376,1428.
- 8 Yi JH Chen Y Li BG et al. Studies on the chemical constituents of the tubers of Curcuma longa. Nat Prod Res Dev(天然产物研究与开发) 2003,15:98-100.
- 9 Guo SM Fan XW Song SG Extraction isolation and identification of ferulic acid from Zizyphus spinosus Hu. Northwest Pharm J ,1995 ,10(1): 22-23.
- 10 Wang XM ,Zhang Q ,Rena K ,et al. Study on the chemical constituents of Cynomorium songaricum. Chin Tradit Herb Drugs 2011 42: 458-460.
- 11 Hu XJ Jin HZ Su J et al. Chemical constituents from Daphne koreana Nakai. Chin J Nat Med 2008 6:411-414.
- 12 Peng T Qiu JP Deng Y et al. Chemical constituents of Primula sikkmensis Hook(II). Nat Prod Res Dev(天然产物研究与开发) 2012 24:1385-1386 1401.
- 13 Fen ZM Li FS Xu JF et al. Chemical constituents from roots of Mallotus apelta. Chin Tradit Herb Drugs 2012 A3: 1489– 1491.